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Key indicators

Single-crystal X-ray study T = 293 K Mean σ (C–C) = 0.007 Å R factor = 0.070 wR factor = 0.214 Data-to-parameter ratio = 16.4

For details of how these key indicators were automatically derived from the article, see http://journals.iucr.org/e. The title compound, $C_8H_9Cl_2N_3$, was synthesized by the reaction of ethyl acetimidate hydrochloride and 2,4-dichlorophenylhydrazine. Both intramolecular $N-H\cdots Cl$ and intermolecular $N-H\cdots N$ hydrogen-bond interactions are present, resulting in a one-dimensional supramolecular chain structure.

 N^{1} -(2,4-Dichloroanilino)acetamidine

Comment

1-Aryl-1,2,4-triazolin-5-one derivatives exhibit herbicidal activity (Mitsuru *et al.*, 1982), for example, with soybeans (Keifer & Tymonko, 1990), by preventing or destroying undesired plant growth (Theodoridis, 1991). The title compound, (I), is an important intermediate in the preparation of triazolinones (Goudar, 2003) and shows good photoconductivity (Chi *et al.*, 2001). We report here the crystal structure of the title compound, (I).



In the molecule of (I), all bond lengths and angles are in normal ranges. The molecular structure of (I) is shown in Fig. 1, where the dashed line indicates an intramolecular $N-H\cdots$ Cl hydrogen bond (Table 1). The dihedral angle between the ring plane and the N2/C7/C8/N3 plane is 67.0 (2)°.

In the crystal structure, intermolecular $N-H\cdots N$ hydrogen bonds (Fig. 2 and Table 1) link the molecules into onedimensional supramolecular chains.

Experimental

A stirred solution of 13.6 g (0.11 mol) of ethyl acetimidate hydrochloride and 13.2 g (0.13 mol) of triethylamine in 100 g of dichloromethane was cooled to about 273 K for 5 min and 17.7 g (0.10 mol) of 2,4-dichlorophenylhydrazine was added. When the addition was complete, the reaction mixture was stirred at 273 K for about 1 h; it was then allowed to warm to ambient temperature and stirred for about 2 h. The reaction mixture was then washed with about 20 ml

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water and dried with magnesium sulfate. The mixture was filtered and the filtrate was concentrated under reduced pressure. The residue was triturated with 20 ml of hexane, and the resultant solid was collected by filtration. The solid was washed with 50 ml of hexane and dried, yielding 20.4 g of 2-(2,4-dichlorophenyl)hydrazidethaneimidic acid. The compound was crystallized twice by slow evaporation of an ethanol solution and crystals suitable for X-ray diffraction were obtained.

Crystal data

$C_8H_9Cl_2N_3$
$M_r = 218.08$
Monoclinic, C2/c
a = 22.832(5) Å
b = 4.756 (1) Å
c = 20.515 (4) Å
$\beta = 115.74 \ (3)^{\circ}$
$V = 2006.7 (9) \text{ Å}^3$

Data collection

Enraf-Nonius CAD-4 diffractometer $\omega/2\theta$ scans Absorption correction: ψ scan (North *et al.*, 1968) $T_{\rm min} = 0.795, T_{\rm max} = 0.942$ 2013 measured reflections

Refinement

Refinement on F^2 $R[F^2 > 2\sigma(F^2)] = 0.070$ $wR(F^2) = 0.214$ S = 1.041966 reflections 120 parameters H-atom parameters constrained Z = 8 $D_x = 1.444 \text{ Mg m}^{-3}$ Mo K α radiation $\mu = 0.60 \text{ mm}^{-1}$ T = 293 (2) K Plate, colourless $0.40 \times 0.30 \times 0.10 \text{ mm}$

1966 independent reflections 1178 reflections with $I > 2\sigma(I)$ $R_{int} = 0.026$ $\theta_{max} = 26.0^{\circ}$ 3 standard reflections every 200 reflections intensity decay: none

 $w = 1/[\sigma^{2}(F_{o}^{2}) + (0.0946P)^{2} + 3.9518P]$ where $P = (F_{o}^{2} + 2F_{c}^{2})/3$ $(\Delta/\sigma)_{max} = 0.027$ $\Delta\rho_{max} = 0.38 \text{ e } \text{\AA}^{-3}$ $\Delta\rho_{min} = -0.33 \text{ e } \text{\AA}^{-3}$ Extinction correction: SHELXL97 Extinction coefficient: 0.0105 (16)

Table 1 Hydrogen bond geometr

Hydrogen-bond geometry (Å, °).

$D - H \cdots A$	D-H	$H \cdot \cdot \cdot A$	$D \cdots A$	$D - H \cdots A$
N1-H1···Cl1	0.90	2.53	2.963 (4)	110
$N2-H2\cdots N3^i$	0.86	2.13	2.788 (5)	134

Symmetry code: (i) x, y - 1, z.

The H atoms attached to N were located in a difference map and then refined as riding (N-H = 0.83-0.90 Å). Carbon-bound H atoms were placed in idealized positions and refined as riding, with C-H = 0.93-0.96 Å; for all H atoms, $U_{iso}(H) = 1.2$ or 1.5 times U_{eq} (carrier atom).

Data collection: *CAD-4 Software* (Enraf–Nonius, 1989); cell refinement: *CAD-4 Software*; data reduction: *XCAD4* (Harms & Wocadlo,1995); program(s) used to solve structure: *SHELXS97* (Sheldrick, 1997); program(s) used to refine structure: *SHELXL97*



Figure 1

A view of the molecular structure, with the atom-numbering scheme. Displacement ellipsoids are drawn at the 30% probability level. The dashed line indicates a hydrogen bond.



Figure 2

A partial packing diagram of (I). Hydrogen bonds are shown as dashed lines.

(Sheldrick, 1997); molecular graphics: *SHELXTL* (Siemens, 1996); software used to prepare material for publication: *SHELXL*.

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